

UČNI NAČRT PREDMETA / COURSE SYLLABUS

Ime predmeta: **Kemija**
Course title: **Chemistry**

Študijski program in stopnja Study programme and cycle	Študijska smer Study option	Letnik Year of study	Semester Semester
Splošna medicina, enovit magistrski študijski program		Prvi	1.
General medicine, Uniform master's degree study program		First	1st

Vrsta predmeta (obvezni ali izbirni) /
Course type (compulsory or elective)

obvezni

compulsory

Univerzitetna koda predmeta / University course code:

Predavanja Lectures	Seminar Seminar	Vaje Tutorial	Klinične vaje Clinical training	Druge oblike študija Other forms of study	Samost. delo Individual work	ECTS
30	0	AV LV RV 30			30	3

Nosilec predmeta / Course coordinator:

Doc. dr. Maša Knez Hrnčič

Jeziki /Languages:

Predavanja / Lectures:

slovenski/slovene

Vaje / Tutorial:

slovenski/slovene

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Prerequisites for enrolling in the course or for performing study obligations:

Vsebina (kratek pregled učnega načrta):

- Zgradba atoma, kemične vezi, medmolekulske sile, biološko pomembni elementi, radioizotopi.
- Voda: strukture, lastnosti, H-vezi, hidrofobne interakcije, voda kot topilo.
- Raztopine: raztapljanje plinov v vodi, koligativne lastnosti raztopin, osmozni pojavi v celici, osmodiuretiki.
- pH: ionizacija vode, Kw, pH, šibki in močni elektroliti, kisline in baze, pufri, puferski sistemi v organizmu, porazdelitev ionov v organizmu, biološki pomen pH.
- Oksidoredukcija: definicije, kvantitativna karakterizacija redoks reakcij.
- Redoks potencial in reakcijska prosta entalpija.
- Hitrost kemičnih reakcij: definicije, red in molekularnost reakcij.
- Hitrost kemičnih reakcij in ravnotežje.

Content (syllabus outline):

- Structure of atom, chemical bound and intermolecular forces, biologically important elements, radioisotops.
- Water: structures, properties, H-bound, hydrophobic interactions, water as solvent.
- Solutions: solubility of gases in water, colligative properties of solutions, osmotic phenomenon in the cell, osmo-diuretics agents.
- pH: ionization of water, Kw, pH, weak and strong electrolytes, acids and bases, buffers, buffer systems in organism, distribution of ions in the body, biological importance of pH.
- Oxidoreduction: definition, quantitative characterization of redox reactions.
- Redox potential and reaction free enthalpy.
- Kinetics of chemical reactions: definitions, order and molecularity of reactions.

• Vpliv koncentracije, pH, ionske moči in temperature na hitrost reakcije. • Molekulske osnove življenja: biološko pomembni elementi, ioni in biomolekule. • Organske biomolekule: izomerija, medsebojni vpliv funkcionalnih skupin. • Pregled organskih spojin po funkcionalnih skupinah. • Biološko pomembne organske molekule s primeri v medicini. • Kemija ogljikovih hidratov: monosaharidi, disaharidi, polisaharidi, homoglikani in heteroglikani. • Kemija lipidov in steroidov. • Aminokisline. • Nukleotidi in nukleinske kisline. • Vitamini. • Molekulsko modeliranje medicinsko pomembnih molekul in zdravil.	• Kinetics and equilibrium of chemical reactions. • Influence of concentration, pH, ionic power and temperature on chemical reaction. • Molecular basics of life: biological important elements, ions and biomolecules. • Organic biomolecules: isometry, interacting influence of functional groups. • Review of organic substances according to their functional groups. • Biologically important organic molecules with examples in medicine. • Chemistry of carbohydrates: monosaccharides, disaccharides, polysaccharides, homoglycanes and heteroglycanes. • Chemistry of lipids and steroids. • Amino acids. • Nucleotides in nucleonic acids. • Vitamines. • Molecular modeling of medically important molecules and medicines
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Temeljni literatura in viri / Reading materials:

Temeljna literatura

- F. Lazarini, J. Brenčič: Splošna in anorganska kemija, Fakulteta za kemijo in kemijsko tehnologijo, Ljubljana, 2014.
- A. Zeeck, S. Grond, S. C. Zeeck: Chemie für Mediziner, Published by Urban & Fischer/Elsevier, 2017.
- P. M. Dewick: Essentials of Organic Chemistry: For Students of Pharmacy, Medicinal Chemistry and Biological Chemistry, John Wiley & Sons, 2013.

Dopolnilna literatura

- A. L. Lehninger, D. L. Nelson, M. M. Cox,: Principles of biochemistry 5 th ed., 3rd printing, Worth, New York, 2008.
- D. D. Ebbing, S. D. Gammon: General chemistry, 9th ed., Houghton Mifflin, cop., Boston, New York, 2009.
- Trevor Palmer, Philip L.R. Bonner, Enzymes: Biochemistry, Biotechnology, Clinical Chemistry, Woodhead Publishing 2007.

Cilji in kompetence:

Cilj tega predmeta je obnoviti osnovna znanja iz splošne kemije in poznavanja kemijske zgradbe molekul in reakcij, ter razumeti kemijske reakcije in procese v človeškem organizmu.

Objectives and competences:

The objective of this course is to renew the basic knowledge of general chemistry and knowledge of the chemical structure of molecules and reactions, and to understand the chemical reactions and processes in human body.

Predvideni študijski rezultati:

Znanje in razumevanje:

Po zaključku tega predmeta bo študent sposoben:

- prepoznati in razlikovati molekule,
 - razumeti kemijske reakcije, ki potekajo v človeškem organizmu,
- razložiti transportne pojave v človeškem organizmu.

Prenesljive/ključne spretnosti in drugi atributi:

- delo v skupini,

Intended learning outcomes:

Knowledge and understanding:

On completion of this course the student will be able:

- to recognize and differentiate molecules,
- to understand the chemical reactions in human body,
- to explain transport phenomenon in human body

Transferable/Key Skills and other attributes:

- team work,

spretnost računanja.	computation skill.
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Metode poučevanja in učenja:

- predavanja,
- laboratorijske vaje.

Learning and teaching methods:

- lectures,
- lab work.

Načini ocenjevanja:

Delež (v %) /
Share (in %)

Assessment methods:

Način (pisni izpit, ustno izpraševanje, naloge, projekt)		Type (examination, oral, coursework, project):
• pisni izpit • ustni izpit • opravljene laboratorijske vaje		• written examination • oral examination • completed lab work
ŠTUDIJSKE OBVEZNOSTI ŠTUDENTOV	60 %	ACADEMIC OBLIGATIONS OF STUDENTS:
opravljene laboratorijske vaje, ustni in pisni izpit	30 %	completed laboratory work, oral and written exam
	10 %	
POGOJI ZA PRISTOP K POSAMEZNEMU PREVERJANJU ZNANJA		REQUIREMENTS FOR ACCESS TO INDIVIDUAL KNOWLEDGE CHECKING:
Uspešno opravljen praktični del laboratorijskih vaj je pogoj za pristop na kolokvij, ki je pogoj za pristop na pisni izpit. Kolokvij iz laboratorijskih vaj je pisno preverjanje opravljenih laboratorijskih vaj.		Successfully completed practical part of laboratory work is a requirement for access to the partial exam which is a requirement for access to the written exam. Partial exam in laboratory work is written checking of completed laboratory work.
Ocena vaj pa je sestavljena iz sodelovanja na vajah, kar je ovrednoteno z 20% ocene, ostalih 80% predstavlja kolokvij iz vaj.		Assessment of laboratory work consists of cooperation during laboratory work which is evaluated with 20% of the mark and the rest 80% of the mark is represented by the partial exam in laboratory work.
Pozitivno opravljen pisni del izpita je pogoj za opravljanje ustnega dela izpita. Izpit je pisni in ustni in je sestavljen iz nalog iz vseh področij. Tudi ustni del izpita se opravlja praviloma pisno.		Successfully completed written part of the exam is a requirement for taking the oral part of the exam. The exam is written and oral and consists of tasks from all areas. Also the oral part of the exam is regularly performed in a written form. The exam mark consists of marks of individual areas of the written exam and of individual areas of the oral exam.
Ocena izpita je sestavljena iz ocen posameznih področij pisnega izpita in posameznih področij ustnega izpita.		

Reference nosilca / Course coordinator's references:

MIRKOVIĆ, Dušica, IBRIĆ, Svetlana, BALANČ, Bojana D., KNEZ, Željko, BUGARSKI, Branko M. Evaluation of the impact of critical quality attributes and critical process parameters on quality and stability of parenteral nutrition nanoemulsions. Journal of drug delivery science and technology, ISSN 1773-2247, 2017, vol. 39, str. 341-347, ilustr., doi: 10.1016/j.jddst.2017.04.004. [COBISS.SI-ID 20561430], [JCR, SNIP, WoS do 14. 7. 2017: št. citatov (TC): 0, čistih citatov (CI): 0, čistih citatov na avtorja (CIAu): 0, Scopus do 12. 6. 2017: št. citatov (TC): 0, čistih citatov (CI): 0, čistih citatov na avtorja (CIAu): 0]

KOTNIK, Petra, KOREN KRAJNC, Metka, PAHOR, Artur, FINŠGAR, Matjaž, KNEZ, Željko. HPLC-MS/MS method optimisation for matrix metalloproteinase 3 and matrix metalloproteinase 9 determination in human blood serum using target analysis. *Journal of pharmaceutical and biomedical analysis*, ISSN 0731-7085. [Print ed.], 20 Feb. 2018, vol. 150, str. 137-143, doi: 10.1016/j.jpba.2017.11.067. [COBISS.SI-ID 20988694], [JCR, SNIP, WoS do 23. 2. 2018: št. citatov (TC): 0, čistih citatov (CI): 0, čistih citatov na avtorja (CIAu): 0, Scopus do 23. 12. 2017: št. citatov (TC): 0, čistih citatov (CI): 0, čistih citatov na avtorja (CIAu): 0]

KRAVANJA, Gregor, KNEZ, Željko, KOTNIK, Petra, LJUBEC, Barbara, KNEZ HRNČIČ, Maša. Formulation of nimodipine, fenofibrate, and o-vanillin with Brij S100 and PEG 4000 using the PGSS process. *The Journal of supercritical fluids*, ISSN 0896-8446. [Print ed.], May 2018, vol. 135, str. 245-253, doi: 10.1016/j.supflu.2018.01.021. [COBISS.SI-ID 21211414], [JCR, SNIP, WoS do 15. 10. 2018: št. citatov (TC): 2, čistih citatov (CI): 1, čistih citatov na avtorja (CIAu): 0.20, Scopus do 29. 8. 2018: št. citatov (TC): 2, čistih citatov (CI): 0, čistih citatov na avtorja (CIAu): 0] kategorija: 1A1 (Z, A', A1/2); uvrstitev: SCI, Scopus, MBP; tip dela je verificiral OSICT točke: 20.05, št. avtorjev: 5

KNEZ HRNČIČ, Maša, CÖR, Darija, KNEZ, Željko. Subcritical extraction of oil from black and white chia seeds with n-propane and comparison with conventional techniques. *The Journal of supercritical fluids*, ISSN 0896-8446. [Print ed.], October 2018, vol. 140, str. 182-187, doi: 10.1016/j.supflu.2018.06.017. [COBISS.SI-ID 21526550], [JCR, SNIP, WoS do 19. 10. 2018: št. citatov (TC): 0, čistih citatov (CI): 0, čistih citatov na avtorja (CIAu): 0, Scopus do 14. 7. 2018: št. citatov (TC): 0, čistih citatov (CI): 0, čistih citatov na avtorja (CIAu): 0] kategorija: 1A1 (Z, A', A1/2); uvrstitev: SCI, Scopus, MBP; tip dela je verificiral OSICT točke: 33.41, št. avtorjev: 3

KRAVANJA, Gregor, KNEZ HRNČIČ, Maša, ŠKERGET, Mojca, KNEZ, Željko. Interfacial tension and gas solubility of molten polymer polyethyleneglycol in contact with supercritical carbon dioxide and argon. *The Journal of supercritical fluids*, ISSN 0896-8446. [Print ed.], Feb. 2016, vol. 108, str. 45-55, doi: 10.1016/j.supflu.2015.10.013. [COBISS.SI-ID 19233814], [JCR, SNIP, WoS do 24. 9. 2017: št. citatov (TC): 4, čistih citatov (CI): 2, čistih citatov na avtorja (CIAu): 0.50, Scopus do 31. 8. 2017: št. citatov (TC): 4, čistih citatov (CI): 2, čistih citatov na avtorja (CIAu): 0.50]

BRGLEZ MOJZER, Eva, KNEZ HRNČIČ, Maša, ŠKERGET, Mojca, KNEZ, Željko, BREN, Urban. Polyphenols : extraction methods, antioxidative action, bioavailability and anticarcinogenic effects. *Molecules*, ISSN 1420-3049, 2016, vol. 21, no. 7, str. 1-38. <http://www.mdpi.com/1420-3049/21/7/901>, doi: 10.3390/molecules21070901. [COBISS.SI-ID 19672598], [JCR, SNIP, WoS do 24. 9. 2017: št. citatov (TC): 8, čistih citatov (CI): 8, čistih citatov na avtorja (CIAu): 1.60, Scopus do 30. 9. 2017: št. citatov (TC): 16, čistih citatov (CI): 16, čistih citatov na avtorja (CIAu): 3.20]